

## High pressure study of Pentaerythritol: a synchrotron infrared study

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Pentaerythritol (PET) is a simple molecule, which crystallizes as a solid in tetragonal  $I4$  space group under ambient condition. The crystal has a layered structure with the PET molecules in the a-b plane connected by hydrogen bonding whereas the interlayer coupling is through weak Van der Waals force. Here we report a synchrotron radiation infrared high-pressure study upto a pressure of  $\approx 11$  GPa. The high-pressure experiments were carried out on 6% PET mixed with KBr powder in a membrane DAC with 450  $\mu\text{m}$  culet. The IR radiation from the Mirage beamline at LURE synchrotron was focused on a 100  $\mu\text{m}$  x 100  $\mu\text{m}$  spot in the DAC using a NicPlan IR microscope coupled to a FTIR spectrometer (Thermo Nicolet Magma 560). The figure shows the changes in the IR spectra with increasing pressure over the ranges 650 – 600  $\text{cm}^{-1}$  and 2300 – 4000  $\text{cm}^{-1}$  covering the internal modes of PET. The qualitative features of the spectra remain unchanged till  $\approx 4.7$  GPa beyond which the C-C skeletal stretch mode at 1131  $\text{cm}^{-1}$  and the O-H deformation mode at 1410  $\text{cm}^{-1}$  split into two modes. The changes in the frequencies of the bending, twist, deformation etc modes with increasing pressure shows that the modes at 662  $\text{cm}^{-1}$ , 1375  $\text{cm}^{-1}$ , 1384  $\text{cm}^{-1}$  and 1410  $\text{cm}^{-1}$  modes exhibit sudden decrease in the slope beyond 4.7 GPa. The O-H mode at 3227  $\text{cm}^{-1}$  shows anomalous shift with the frequency decreasing at the rate of 27  $\text{cm}^{-1}/\text{GPa}$  till 4.7 GPa and then much more slowly with 3.7  $\text{cm}^{-1}/\text{GPa}$ . All these changes occurring at 4.7 GPa indicate that the crystal undergoes a phase transition beyond this pressure. On reducing the pressure, the spectrum changes to the ambient phase below  $\approx 4.6$  GPa showing that the transition is completely reversible. The transition pressure is close to the prediction made based on steric hindrance between the non-bonded hydrogen atoms.

